Amendments to the Claims:

1. (Currently Amended) A hydrophilic superabsorbent polymer composition comprising an absorbent polymer that is the reaction product of:

a) from about 55 to about 99.9 wt.% based on the absorbent polymer of polymerizable unsaturated acid group containing monomers;

b) a first neutralizing agent selected from monovalent hydroxides, monovalent carbonate, or monovalent bicarbonate salts, or mixtures thereof;

c) a second neutralizing agent comprising a multivalent metal hydroxide; [[and]]

d) from about 0.001 to about 5.0 wt.% based on the weight of a) of internal crosslinking agent; and

e) an initiator for initiation of free-radical polymerization;

wherein the absorbent polymer has a degree of neutralization of more than about [[20]] 25%, and from about 20 mole % to about 75 mole % of the unsaturated acid group containing monomers are neutralized with the first neutralizing agent, and from about 5 mole % to about 40 mole % of the unsaturated acid group containing monomers are neutralized with the second neutralizing agent, at a temperature of about 75°C or less, and the absorbent polymer is formed into an absorbent polymer particle which is surface treated with

from about 0.001 to about 5.0 wt.% of surface crosslinking agent applied to the polymer particle surface; and

wherein the hydrophilic superabsorbent polymer composition has an absorption time of about $5+10\ a^2$ minutes or greater, where a is the mean particle size of the superabsorbent

material in millimeters, a liquid capacity of about 15 g/g or greater, a drop penetration value of about 2 seconds or less, and a floatability of about 50% or less.

2. (Previously Presented) The hydrophilic superabsorbent polymer composition of Claim 1 having a liquid capacity of about 20 g/g or greater.

3. (Previously Presented) The hydrophilic superabsorbent polymer composition of Claim 1 having a liquid capacity of about 25 g/g or greater.

4. (Previously Presented) The hydrophilic superabsorbent polymer composition of Claim 1 having an Absorption Time of about 7+10 a^2 minutes or greater.

5. (Previously Presented) The hydrophilic superabsorbent polymer composition of Claim 1 having an Absorption Time of about 10+10 a^2 minutes or greater.

6. (Previously Presented) The hydrophilic superabsorbent polymer composition of Claim 1 having a Gel Bed Permeability of about 20 x 10^{-9} cm^2 or greater.

7. (Previously Presented) The hydrophilic superabsorbent polymer composition of Claim 1 having a Gel Bed Permeability of about 50 x 10^{-9} cm^2 or greater.

8. (Previously Presented) The hydrophilic superabsorbent polymer composition of Claim 1 having a Gel Bed Permeability of about 80 x 10^{-9} cm^2 or greater.

9. (Previously Presented) The hydrophilic superabsorbent polymer composition of Claim 1 wherein the first neutralizing agent is sodium hydroxide, and the second neutralizing agent is selected from calcium hydroxide or magnesium hydroxide.

10. (Previously Presented) The hydrophilic superabsorbent polymer composition of Claim 1 wherein at least 40% of the neutralization is accomplished by the first neutralizing agent.

11. (Previously Presented) The hydrophilic superabsorbent polymer composition of Claim 1 wherein the first neutralizing agent comprises a monovalent metal hydroxide.

12. (Canceled)

13. (Currently Amended) A water insoluble, cross-linked, partially neutralized, hydrophilic, superabsorbent polymer composition having a degree of neutralization of from about 20 mole % to about 75 mole %, wherein the hydrophilic superabsorbent polymer composition comprises an absorbent polymer that is the reaction product of a polymerizable unsaturated acid group containing monomers; an internal crosslinking agent; an initiator for initiation of free-radical polymerization, a first neutralizing agent selected from monovalent hydroxide, monovalent carbonate, or bicarbonate salts, or mixtures thereof; and from about 5 mole % to about 40 mole % of the unsaturated acid group containing monomers are neutralized with a second neutralizing agent comprising a multivalent metal hydroxide, wherein

the hydrophilic superabsorbent polymer composition has an absorption time of about $5+10\ a^2$ minutes or greater, where a is the mean particle size of the superabsorbent material in millimeters, a liquid capacity of about 15 g/g or greater, a drop penetration value of about 2 seconds or less, and a floatability of about 50% or less.

14. (Previously Presented) The water insoluble, cross-linked, partially neutralized, hydrophilic, superabsorbent polymer composition of Claim 13 having a liquid capacity of about 20 g/g or greater.

15. (Previously Presented) The water insoluble, cross-linked, partially neutralized, hydrophilic, superabsorbent polymer composition of Claim 13 having a liquid capacity of about 25 g/g or greater.

16. (Previously Presented) The water insoluble, cross-linked, partially neutralized, hydrophilic, superabsorbent polymer composition of Claim 13 having an Absorption Time of about $7+10\ a^2$ minutes or greater.

17. (Previously Presented) The water insoluble, cross-linked, partially neutralized, hydrophilic, superabsorbent polymer composition of Claim 13 having an Absorption Time of about $10+10\ a^2$ minutes or greater.

18. (Previously Presented) The water insoluble, cross-linked, partially neutralized, hydrophilic, superabsorbent polymer composition of Claim 13 having a Gel Bed Permeability of about 20 x 10^{-9} cm^2 or greater.

19. (Previously Presented) The water insoluble, cross-linked, partially neutralized, hydrophilic, superabsorbent polymer composition of Claim 13 having a Gel Bed Permeability of about 50 x 10^{-9} cm^2 or greater.

20. (Previously Presented) The water insoluble, cross-linked, partially neutralized, hydrophilic, superabsorbent polymer composition of Claim 13 having a Gel Bed Permeability of about 80 x 10^{-9} cm^2 or greater.

21. (Currently Amended) A hydrophilic superabsorbent polymer composition comprising an absorbent polymer that is the reaction product of:

a) from about 55 to about 99.9 wt.% based on the absorbent polymer of polymerizable unsaturated acid group containing monomers;

b) a first neutralizing agent selected from monovalent hydroxides, monovalent carbonate, or bicarbonate salts, or mixtures thereof;

c) a second neutralizing agent comprising a multivalent metal hydroxide; [[and]]

d) from about 0.001 to about 5.0 wt.% based on the weight of a) of internal crosslinking agent; and

e) an initiator for initiation of free-radical polymerization;

wherein the absorbent polymer has a degree of neutralization of more than about [[20]]25%, and from about 20 mole % to about 75 mole % of the unsaturated acid group containing monomers are neutralized with the first neutralizing agent, and from about 5 mole % to about 40 mole % of the unsaturated acid group containing monomers are neutralized with the second neutralizing agent, and the absorbent polymer is formed into a absorbent polymer particle which is surface treated with

from about 0.001 to about 5.0 wt.% of surface crosslinking agent applied to the particle surface.

22. (Canceled)

23. (Previously Presented) The hydrophilic superabsorbent polymer composition of Claim 21 wherein at least 40% of the neutralization is accomplished by the first neutralizing agent.

24. (Previously Presented) The hydrophilic superabsorbent polymer composition of Claim 21 wherein the first neutralizing agent comprises a sodium hydroxide, and the second neutralizing agent is selected from calcium hydroxide or magnesium hydroxide.

25. (Canceled)